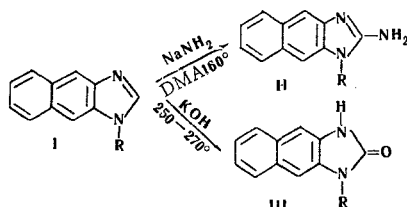


NAPHTHO[2,3-d]IMIDAZOLE IN THE CHICHIBABIN REACTION

I. S. Kashparov, A. F. Pozharskii, and A. M. Simonov

UDC 547.785.5

We have successfully used naphtho[2,3-d]imidazole (I) in the Chichibabin reaction. The amination of I with a twofold excess of  $\text{NaNH}_2$  by the usual method [1,2] takes place readily at  $160^\circ\text{C}$  and is accompanied by the evolution of hydrogen, which continues for 30-40 min. The reaction products, in good yields, are the previously unknown 2-amino-1-R-naphtho[2,3-d]imidazoles (II).



Thus, the amination of all three isomeric naphthimidazoles has been effected (for the amination of the angular and peri isomers, see [1-3]).

Compounds I are also smoothly hydroxylated by solid alkali [4], forming the corresponding naphtho[2,3-d]imidazoles (III).

Compound Ia was obtained by the methylation of I (R = H) [5] with methyl iodide (2 moles) in liquid ammonia in the presence of sodium amide (2 moles); the reaction product was isolated from the reaction mixture by chromatography ( $\text{Al}_2\text{O}_3$  chloroform, the Ia was eluted first). Compound Ib was prepared by the benzylation of I (R = H) with benzyldimethylphenylammonium chloride (2 moles) in aqueous alkali (2 moles) [6]. Other methods for the methylation and benzylation of I (R = H) were unsuccessful.

IR spectra (UR-20, paraffin oil) showed that compound II exists in the solid state mainly in the amino form ( $\nu_{\text{NH}_2}^{\text{as}} 3445 \text{ cm}^{-1}$ ,  $\nu_{\text{NH}_2}^{\text{s}} 3345 \text{ cm}^{-1}$ ) and III in the keto form ( $\nu_{\text{C}=\text{O}} 1710 \text{ cm}^{-1}$ ). The other properties of the compounds (all of which are colorless crystalline substances) are given below. In all cases, the results of their analysis for C, H, and N agreed with the calculated values.

Compound	R	mp, °C	Solvent for recrystallization	Yield, %
Ia	$\text{CH}_3$	162-163	Octane	55
Ib	$\text{C}_6\text{H}_5\text{CH}_2$	209-210	Benzene	67
IIa	$\text{CH}_3$	286-287	Aqueous ethanol	52
IIb	$\text{C}_6\text{H}_5\text{CH}_2$	305	Ethanol	57
III	$\text{C}_6\text{H}_5\text{CH}_2$	297	Ethanol	80

LITERATURE CITED

1. A. D. Garnovskii and A. M. Simonov, *ZhOKh*, **31**, 1941 (1961).
2. B. I. Khristich and A. M. Simonov, *KhGS [Chemistry of Heterocyclic Compounds]*, **2**, 611 (1966).
3. A. F. Pozharskii, I. S. Kashparov, and A. M. Simonov, *KhGS [Chemistry of Heterocyclic Compounds]*, **4**, 183 (1968).
4. I. S. Kashparov and A. F. Pozharskii, *KhGS [Chemistry of Heterocyclic Compounds]*, **6**, (1970).
5. K. Fries, R. Walter, and K. Schilling, *Lieb. Ann.*, **516**, 248 (1935).
6. A. M. Simonov, N. D. Vitkevich, and B. K. Martsokha, *ZhOKh*, **30**, 3063, (1960); *KhGS [Chemistry of Heterocyclic Compounds]*, **6**, 1147, 1970.

Rostov-on-Don State University. Translated from *Khimiya Geterotsiklicheskikh Soedinenii*, Vol. 6, No. 8, pp. 1147-1148, August, 1970. Original article submitted January 27, 1970.

© 1973 Consultants Bureau, a division of Plenum Publishing Corporation, 227 West 17th Street, New York, N. Y. 10011. All rights reserved. This article cannot be reproduced for any purpose whatsoever without permission of the publisher. A copy of this article is available from the publisher for \$15.00.